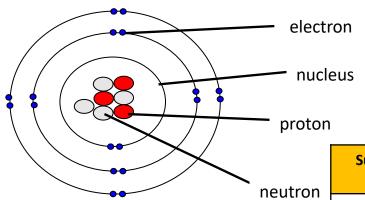
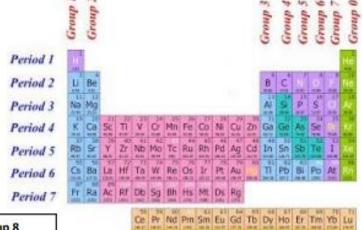
Atoms knowledge organiser



Atoms are the building blocks of all substances. They are made up of sub-atomic particles called **protons**, **electrons and neutrons**.

Sub-atomic particle	Charge	Relative mass	
Proton	Positive (+1)	1	
Electron	Negative (-1)	0.0005	
Neutron	Neutral (0)	1	

Key word	Definition			
Atom	Smallest particle of an element			
Electron	Negatively charged subatomic particle			
Neutron	Neutral subatomic particle			
Nucleus	Central part of an atom, the nucleus containing			
	neutrons and protons			
Proton	on Positively charged subatomic particle			
Shell	Orbit followed by electrons			
Element	A substance made of one type of atoms			
Compound	A substance made of two or more types of atoms			



Group 1	Group 2	Group 3	Group 4	Group 5	Group 6	Group 7	Group 8
Lithium - Li Sodium - Na Potassium - K	Beryllium – Be Magnesium – Mg Calcium - Ca	Boron – B Aluminium – Al Gallium – Ga	Carbon – C Silicon – Si Germanium – Ge	Nitrogen – N Phosphorus – P Arsenic – As	Oxygen – O Sulfur – S Selenium - S	Fluorine – F Chlorine – Cl Bromine - Br	Helium – He Neon – Ne Argon - Ar
(E)	Be	В	C	N	***	* F	X X Ne